LETTERS TO THE EDITOR

SYNTHESIS OF 2,3-DIHYDRO-3H-PYRANO[2,3-f]-1,3-BENZOXAZINES

M. V. Vovk and A. V. Bol'but

A simple synthetic approach to previously unknown angular pyranobenzoxazines consists of cyclocondensation of 7-hydroxycoumarins (Ia, b) with chlorocarbonyl isocyanate [1, 2]. Heating these reagents in boiling chlorobenzene for 2 h results in 2,3-dihydro-3H-pyrano[2,3-f]-1,3-benzoxazine derivatives (IIa, b) with a good yield. In all probability, this conversion takes place according to the scheme of the reaction of chlorocarbonyl isocyanate with phenols [2], while the formation of systems of the angular type is probably the result of the highest susceptibility of position 8 in the coumarin ring to electrophilic attack [3].

$$R^{2}$$
 R^{1}
 R^{2}
 R^{1}
 R^{2}
 R^{2}
 R^{2}
 R^{1}
 R^{2}
 R^{2}
 R^{2}
 R^{1}
 R^{2}
 R^{2}
 R^{2}
 R^{1}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{4}
 R^{4}
 R^{4}
 R^{5}
 R^{4}
 R^{5}
 R^{5

A mixture of 0.01 mole of 7-hydroxycoumarin Ia, b and 1.05 g (0.01 mole) of chlorocarbonyl isocyanate in 25 ml of chlorobenzene was boiled for 2 h, cooled, and the sediment formed was filtered off and crystallized from dioxane-DMF mixture (5:1).

8-Methyl-2,4,6-trioxo-3H-pyrano[2,3-f]-1,3-benzoxazine (IIa). mp = 171-173 °C. IR spectrum (KBr): 1730, 1810 (C=O), 3230 cm⁻¹ (N-H). PMR spectrum (in DMSO-D₆): 2.35 (3H, s, CH₃); 6.12 (1H, s, CH=); 7.44 (1H, d, J=7 Hz, 6-H); 7.88 (1H, d, J=7 Hz, 5-H); 11.94 (1H, s, NH). Yield of 64%. Found, %: C 59.04; H 2.72; N 5.75. $C_{12}H_7NO_5$. Calculated, %: C 58.78; H 2.88; N 5.71.

7,8-Trimethylene-2,4,6-trioxo-3H-pyrano[2,3-f]-1,3-benzoxazine (IIb). mp = 220-221 °C. IR spectrum (KBr): 1735, 1810 (C=O), 3235 cm⁻¹ (N-H). PMR spectrum (in DMSO-D): 2.12-3.04 (6H, m, 3CH₂); 7.35 (1H, d, J=7 Hz, 6-H); 7.62 (1H, d, J=7 Hz, 5-H); 11.73 (1H, s, NH). Yield of 69%. Found, %: C 62.33; H 3.40; N 4.94. $C_{14}H_9NO_5$. Calculated, %: C 62.00; H 3.32; N 5.16.

REFERENCES

- 1. A. Kamal, Heterocycles, 31, 1377 (1990).
- 2. V. I. Gorbatenko, Tetrahedron, 49, 3227 (1993).
- 3. D. Barton and W. D. Ollis, Comprehensive Organic Chemistry, Pergamon Press, Oxford, England (1979).

Institute of Organic Chemistry, National Academy of Sciences of Ukraine, Kiev 253660. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 9, p. 1271, September, 1997. Original article submitted February 7, 1997.